Thermal Physics
(Maxwell Distribution Law)
Lecture -7

Dr. Tarun Kumar Dey, Associate professor Department of Physics,

L.S College; BRA Bihar University, Muzzaffarpur

Youtube channel – Tarun Kumar Dey

Online Course Link:

http::/findmementor.com/mentee/view details/tkdeyphy

$$b = m/2kT \tag{25}$$

Using the values of a and b, Maxwell distribution law of velocities can be written in the form.

$$dn_c = n_c dc = 4\pi n \text{ (m/ 2nkT)}^{3/2} e^{-mc^{2/2kT}} c^2 dc$$
 (26)
The quantity

$$f(c) = dn_c / ndc = 4\pi^{-1/2} (m/2kT)^{3/2} c^2 exp(-mc^2/2kT)(27)$$

It is the distribution function of velocities of molecules by velocities.

From kinetic theory of gas, the pressure P exerted by gas is

$$P = 1/3 \text{ mN} / V c^2$$
 (21)

Where c² is mean square velocity, which is defined as

$$c^2 = 1/n \int_0^\infty c^2 dn_c$$

$$= 4 \pi a^3 \int_0^\infty e^{-bc^2} c^4 dc$$
 (22)

$$bc^{2} = X$$
, we get,
 $c^{2} = 2\pi a^{3}/b^{5/2} \int_{0}^{\infty} e^{-X} X^{1/2} dX$

$$= 2\pi a^3 / b^{5/2} \% \pi^{1/2}$$

$$= 3/2b$$

Then, P = mN / 2Vb

But for ideal gas

P = NkT / V

(24)

(23)

Maxwell distribution some times also called Maxwell Boltzmann distribution law in an equilibrium distribution law.

Experimental verification.

Stern's Experiment: In 1947, Stern, Estermann and simpson arranged an experiment for verifying the Maxwell law of velocity distribution.

The experimental arrangement is shown in fig 2. .The apparatus consists of an open vessel having hot gas with a narrow horizontal opening hole .

Cesium is taken as the source of atoms, which is heated in the oven.

A nozzle slit is placed at a distance of 1meter from it,

A thin tungsten wire placed at a distance of 1 meter from the slit S.

It serves as a target. The entire arrangement are along one strictly horizontal line.

The entire arrangement is enclosed in a highly evacuated chamber .

The cesium atoms flow out of the oven through nozzle.

In the absence of gravitational field ,they would strike the target $\,$. However due to gravitational field the atoms travel along parabola .The atoms emerging from nozzle with a velocity horizontally along the $\,$ X - axis will not pass through the $\,$ narrow slit and will not reach the target $\,$.

The atoms emerging from slit at a small angle θ and will pass through another slit and strike the target .

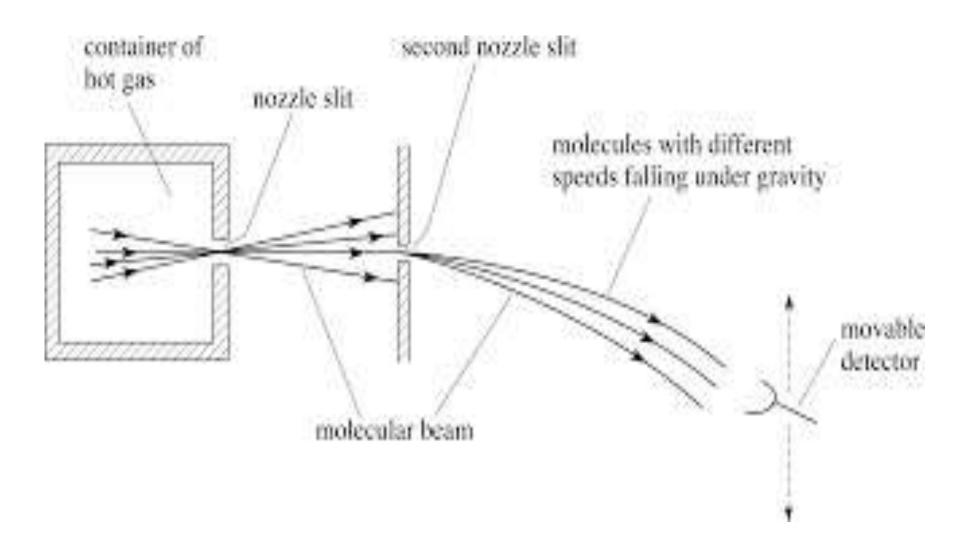


Figure.3 Apparatus of Stern's improved experiment

The tungsten wire —target is heated by an electric current passing through it .

When cesium atoms strike the wire —target, they get ionized. These positively charged ions, leaving the target, get into the negatively charged cylinder surrounding the target.

Thus an electric current of ions passes between the wire and the cylinder, which can be measured with accuracy.

The ionic current gives the number of atoms hitting the target.

Moving the target in a vertical direction at different positions , the ionic current and hence the number of atoms hitting the target is measured at different heights .

We find the number of atoms having different velocities – the atoms hitting target have velocity higher than those at different point . This gives us the distribution of atoms with velocities . This is in complete agreement with the Maxwell distribution law of velocities .