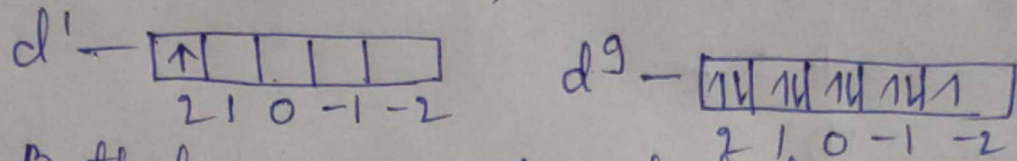


# correlation and spin-orbit coupling in Free ions for First series of Transition Metal :—

1.  $Sc^{2+}$  and  $Cu^{2+}$  ( $d^1, d^9$ )



Both have one unpaired electron.

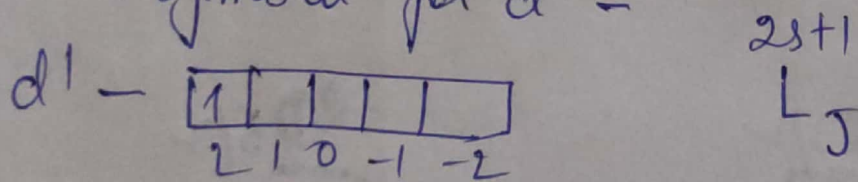
$d^1$  - No. of microstates =  $\binom{n}{r} = \frac{n!}{r!(n-r)!}$

$n = 10, r = 1$   
 $= \binom{10}{1} = \frac{10!}{1!(10-1)!}$

$= \frac{10 \times 9 \times 8 \times 7 \times 6 \times 5 \times 4 \times 3 \times 2 \times 1}{1 \times (9 \times 8 \times 7 \times 6 \times 5 \times 4 \times 3 \times 2 \times 1)}$   
 $= \frac{10}{1} = 10$

$d^9$  - No. of microstates = 10

Term symbol for  $d^1$  —



$S = \frac{1}{2}$

$L = 2$

Term — D

Spin multiplicity —  $2S+1 = 2 \times \frac{1}{2} + 1 = 2$

$J = L+S = 2 + \frac{1}{2} = \frac{5}{2}$

$L+S-1 = \frac{5}{2} - 1 = \frac{3}{2}$

Possible term symbol.

${}^2D_{5/2}, {}^2D_{3/2}$

$$d^9 - \begin{array}{|c|c|c|c|c|} \hline 1 & 1 & 1 & 1 & 1 \\ \hline \end{array}$$

$$2 \quad 1 \quad 0 \quad -1 \quad -2$$

$$S = \frac{1}{2}$$

$$2S+1 = 2$$

$$L = 5$$

$$L = 4 + 2 + 0 - 2 - 2$$

$$= 6 - 4 = 2$$

Term =  $^2D$

spin multiplicity  $2S+1 = 2 \times \frac{1}{2} + 1 = 2$

$$J = L + S = 2 + \frac{1}{2} = \frac{5}{2}$$

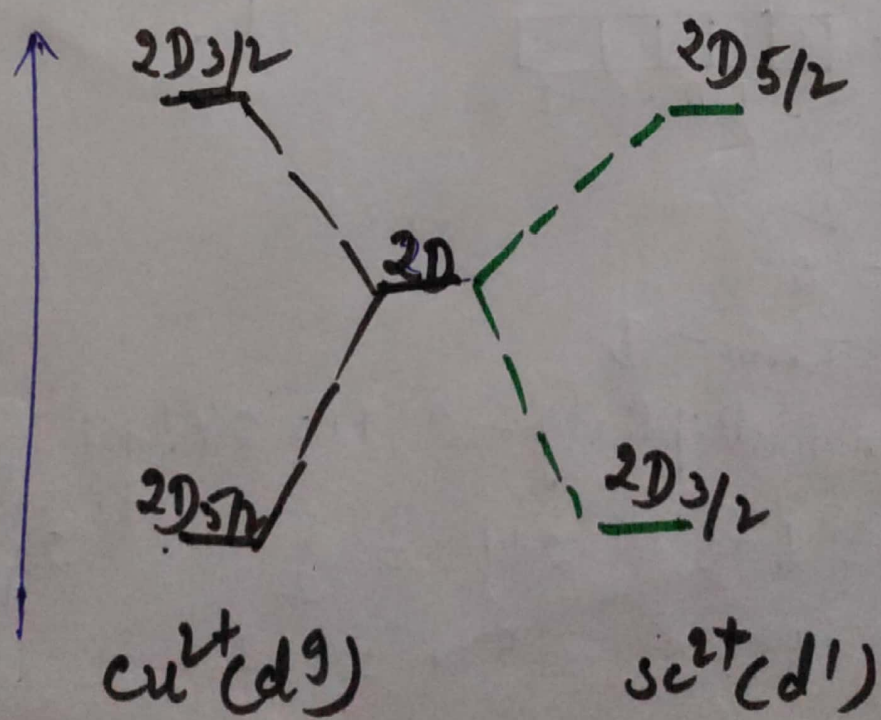
$$L + S - 1 = \frac{3}{2}$$

Possible term symbol =  $^2D_{5/2}, ^2D_{3/2}$

Both ion have same microstate and Term symbol

$^2D$  which split into  $^2D_{5/2}$  &  $^2D_{3/2}$  but

ground state term symbol for  $d^1$  system -  $^2D_{3/2}$  & for  $d^9$  system is  $^2D_{5/2}$ .



correlation and spin-orbital coupling of free ion term of  $Sc^{2+}$  and  $Cu^{2+}$