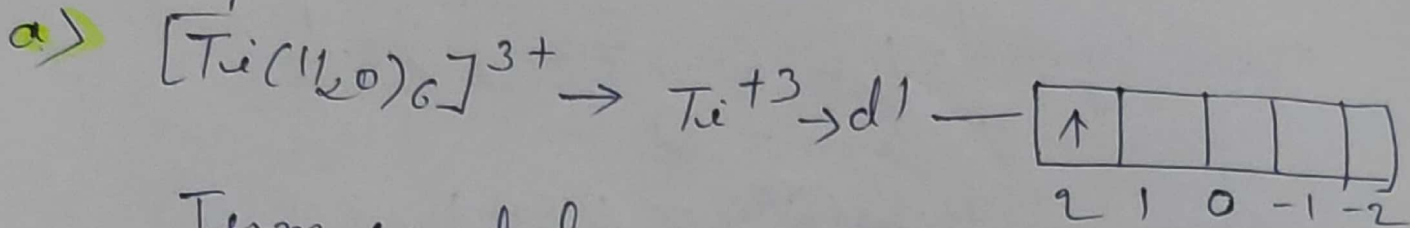


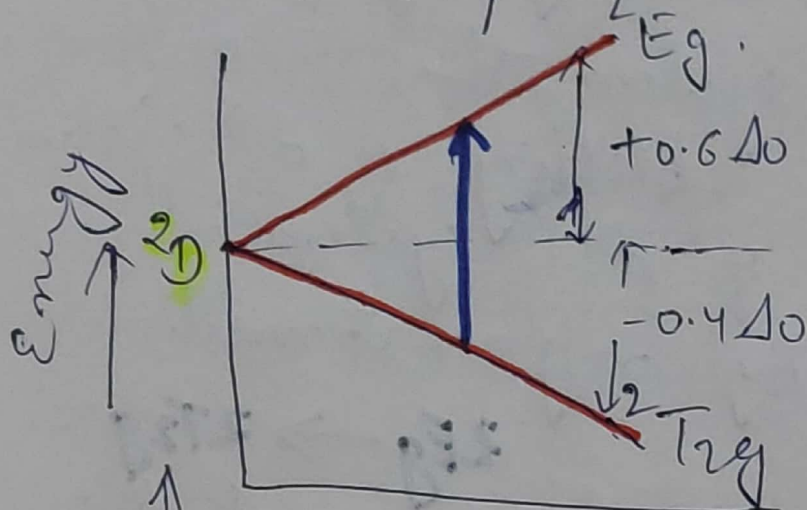
Q Define crystal field diagram and electronic spectra in $[\text{Ti}(\text{H}_2\text{O})_6]^{3+}$ and $[\text{Cu}(\text{H}_2\text{O})_6]^{2+}$ complexes.



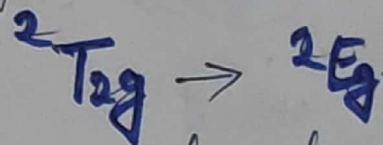
Term symbol — $L=2 \therefore \text{Term} = D$

Spin multiplicity $2S+1 = 2 \times \frac{1}{2} + 1 = 2$
 $= 2D$

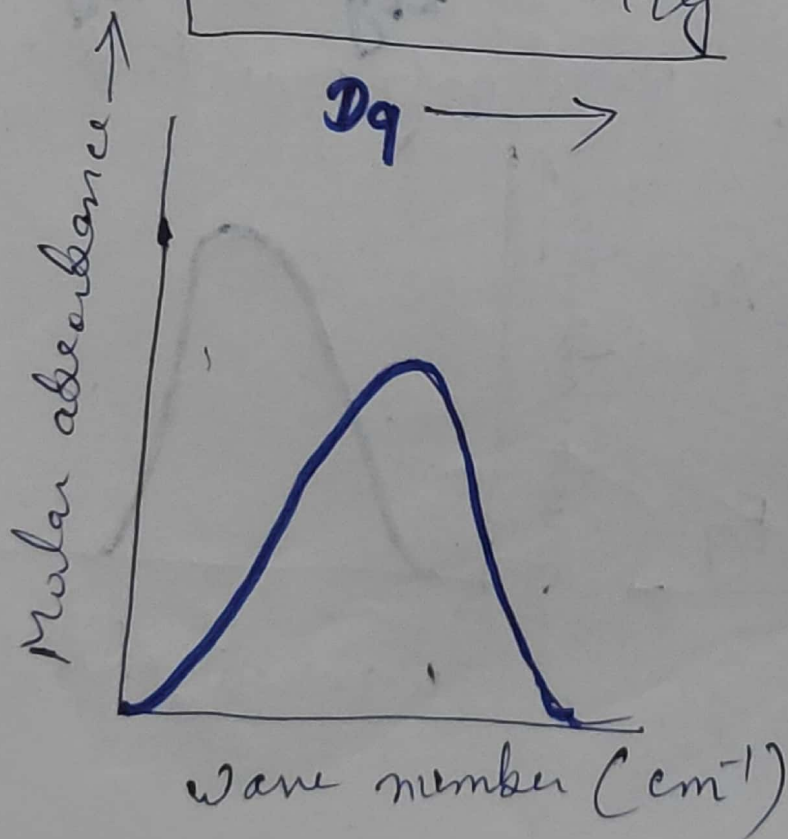
$2D$ Term splits into T_{2g} & E_g .



only one transition



\therefore we get only one peak in spectra.



$[Cu(H_2O)_6]^{2+} \rightarrow Cu^{2+} \rightarrow d^9$ configuration

1	1	1	1	1
2	1	0	-1	-2

Term symbol - $L = 6 - 4 = 2$, $S = \frac{1}{2}$

\therefore Term = D

Spin multiplicity = $2S + 1$
 $= 2 \times \frac{1}{2} + 1 = 2$

Term symbol **2D**

Splitting of 2D \rightarrow ${}^2T_{2g} + {}^2E_g$

Since in d^9 system there is only one whole therefore splitting of d^9 system is just inverse of d^1 system:

